

6,7-Dimethyl-3,4,4a,7,8,8a-hexahydro-
isochromen-1-oneNijhuma Kayal and Vickie
McKee*Chemistry Department, Loughborough
University, Loughborough, Leicestershire
LE11 3TU, England

Correspondence e-mail: v.mckee@lboro.ac.uk

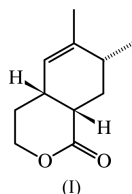
Key indicators

Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.040
 wR factor = 0.100
Data-to-parameter ratio = 14.7For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_{11}\text{H}_{16}\text{O}_2$, crystallized in space group $P2_1/n$ with two very similar molecules in the asymmetric unit. In each molecule, the two rings are fused in the *endo* configuration.

Comment

The title lactone, (I), was prepared as an intermediate in the synthesis of an isomer of the oxaspirobicyclic tetrionic acid unit of the CCK-B receptor antagonist tetronothiodin.



The asymmetric unit of (I) contains two independent molecules (Fig. 1) which are essentially superimposable. The structure determination confirms the expected geometry at atoms C2, C4 and C9 in each molecule and there are no unusual bond lengths or angles.

Fig. 2 shows the unit-cell packing; the molecules are stacked parallel to the *b* axis. There are indications of weak C—H...O hydrogen bonds between neighbouring molecules; these are listed in Table 1 (Desiraju, 1996; Taylor & Kennard, 1982).

Experimental

The title lactone was formed by oxidation of the analogous lactol, using pyridinium dichromate, and recrystallized from dichloromethane solution by evaporation. The synthesis of the compound has been reported elsewhere (Page *et al.*, 2003).

Crystal data

$\text{C}_{11}\text{H}_{16}\text{O}_2$
 $M_r = 180.24$
Monoclinic, $P2_1/n$
 $a = 14.2924$ (17) Å
 $b = 5.4060$ (6) Å
 $c = 25.969$ (3) Å
 $\beta = 96.530$ (2)°
 $V = 1993.4$ (4) Å³
 $Z = 8$

$D_x = 1.201$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 3712
reflections
 $\theta = 2.6$ – 27.8 °
 $\mu = 0.08$ mm⁻¹
 $T = 150$ (2) K
Lath, colourless
 $0.48 \times 0.11 \times 0.03$ mm

Data collection

Bruker SMART CCD area-detector
diffractometer
 φ and ω scans
Absorption correction: multi-scan
(*SHELXTL/SADABS*;
Sheldrick, 1998)
 $T_{\min} = 0.938$, $T_{\max} = 1.000$
13 507 measured reflections

3517 independent reflections
2526 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$
 $\theta_{\text{max}} = 25.0$ °
 $h = -16 \rightarrow 16$
 $k = -6 \rightarrow 6$
 $l = -30 \rightarrow 30$

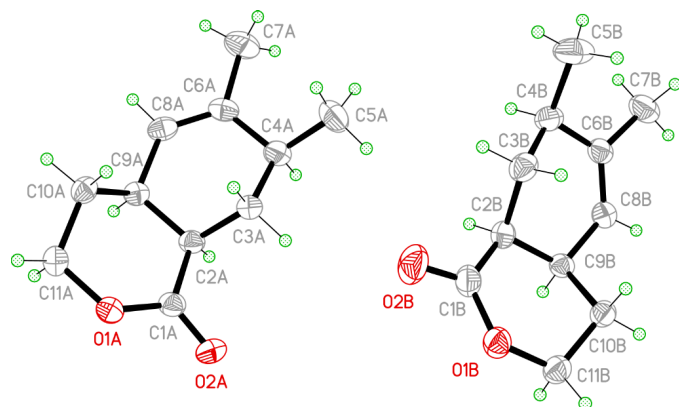


Figure 1
Perspective view of the two independent molecules in the asymmetric unit of (I), showing 50% probability displacement ellipsoids.

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.100$
 $S = 1.04$
 3517 reflections
 239 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0381P)^2 + 0.7308P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.21 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.17 \text{ e } \text{Å}^{-3}$

Table 1

Hydrogen-bonding geometry ($\text{Å}, ^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C8B-H8B \cdots O2A^i$	0.95	2.48	3.304 (2)	144
$C9A-H9A \cdots O1A^{ii}$	1.00	2.80	3.577 (2)	135
$C11A-H11A \cdots O2A^{iii}$	0.99	2.82	3.455 (2)	122
$C11A-H11B \cdots O2A^{iv}$	0.99	2.55	3.510 (2)	162
$C9B-H9B \cdots O1B^i$	1.00	2.88	3.569 (2)	127
$C9B-H9B \cdots O1B^{ii}$	1.00	2.76	3.529 (2)	134
$C11B-H11C \cdots O2B^i$	0.99	2.61	3.567 (2)	164

Symmetry codes: (i) $\frac{3}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$; (ii) $x, y - 1, z$; (iii) $\frac{1}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$; (iv) $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$.

H atoms were introduced at calculated positions; the constrained C—H distances were 0.95, 0.98, 0.99 and 1.00 Å for H atoms bonded

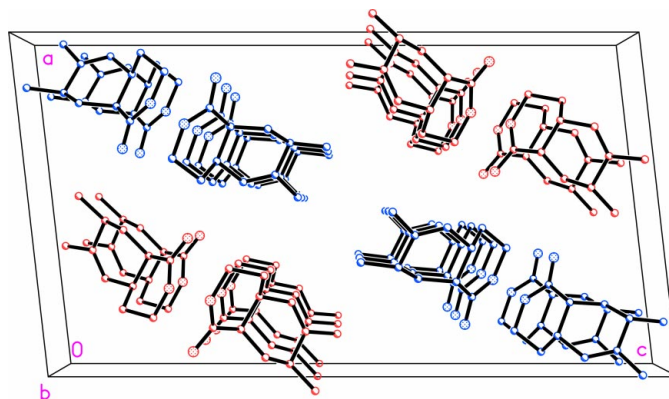


Figure 2
Projection of the unit-cell contents along b . Atoms labelled A in are shown in red, those labelled B are in blue.

to sp^2 , methylene, methyl and tertiary C atoms, respectively. H atoms were refined with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$, except for the methyl groups where $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 1998); software used to prepare material for publication: *SHELXTL*.

We are grateful to Hooshang Vahedi for supplying the crystals.

References

- Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Desiraju, G. R. (1996). *Acc. Chem. Res.* **29**, 441–449.
 Page, P. C. B., Vahedi, H., Batchelor, K. J., Hindley, S. J. & Beswick, P. (2003). *Synlett*. In the press.
 Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
 Sheldrick, G. M. (1998). *SHELXTL/SADABS*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
 Taylor, R. & Kennard, O. (1982). *J. Am. Chem. Soc.* **104**, 5063–5070.